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3 SEM PG (CBCS) CHM AEC 6

2024

(December)

CHEMISTRY

Paper : AEC-306

(Analytical Chemistry-II)

Full Marks : 30

Time : Two hours

The figures in the margin indicate full marks for the questions.

Write the answers of Unit-I and Unit-II in separate books.

UNIT-I

(Marks : 15)

1. Answer **any four** from the following questions : 3×4=12

(a) What are the common adsorbents used in column chromatography? Write the applications of adsorption chromatography. 1+2=3

Contd.

- (b) Briefly describe the basic working principle of gas chromatography with the help of a diagram.
- (c) Write the working principle of paper chromatography. Write the applications and limitations of paper chromatography. $1+1+1=3$
- (d) Write the components of HPLC (High Pressure Liquid Chromatography) mentioning their functions. $2+1=3$
- (e) Write the basic principle of separation using Ion-exchange Chromatography with a schematic diagram.
2. Write a short note on : (*any one*) 3
- (a) Flash chromatography
- (b) Gel permeation chromatography

UNIT-II

(Marks : 15)

3. Answer *any three* of the following : $3 \times 3 = 9$
- (a) Describe the *two* primary types of electron guns used in Transmission Electron Microscopy (TEM).

- (b) Explain the principle of SEM with the help of a diagram.
- (c) What is the major drawback of contact mode AFM? Explain how this issue is addressed by using non-contact or tapping mode AFM. $1+2=3$
- (d) Explain the basic working principle of AFM. Name the different components involved in an AFM system. $2+1=3$
- (e) Describe the roles of various lenses and apertures used in TEM.

4. Answer *any three* of the following : $2 \times 3 = 6$

- (a) What are the limitations of SEM?
- (b) What is the typical energy/wavelength of electrons employed in TEM? Why is it necessary to operate at such high energies? $1+1=2$
- (c) Why is it necessary to coat samples before imaging in SEM?
- (d) Write a short note on sample preparation for TEM.